

# Ch 10 and 11 HW

10-1 Show that for given E, smaller rest mass leads to smaller wavelength.

$$E^2 = p^2 c^2 + E_0^2$$
$$= \left(\frac{h}{\lambda}\right)^2 c^2 + (m_0 c^2)^2$$

Solve  $\lambda^2 = \frac{h^2}{\frac{E^2}{c^2} - \frac{(m_0 c^2)^2}{c^2}}$

$E > m_0 c^2$

So, smaller mass increases the denominator for a fixed Energy E.  
The wavelength reduces.

10-5 Given a neutron,  $1.675 \times 10^{-27}$  kg and wavelength of 100pm, how fast?

$$p = \gamma m_0 v = \frac{h}{\lambda}$$

$$\gamma v = \frac{h}{m_0 \lambda}$$

square

$$\frac{v^2}{1 - v^2/c^2} = \frac{h^2}{m_0^2 \lambda^2}$$

Algebra - solve  $v$

$$v = \frac{h/m_0 \lambda}{\sqrt{1 + (h/m_0 \lambda)^2 \left(\frac{1}{c}\right)^2}}$$

$$\sqrt{1 + (h/m_0 \lambda)^2 \left(\frac{1}{c}\right)^2}$$

$$= 3956 \text{ m/s}$$

Slow, so we could have just ignored gamma. Could have checked energy to start.

10-6 Express wavelength  $\lambda = h/p$  in terms of KE for relativistic various cases (a,b,c, d)

$$E^2 = p^2 c^2 + E_0^2$$

a) photons so  $E_0 \rightarrow 0$   $E \rightarrow KE$

$$\lambda = \frac{h}{p} = \frac{hc}{KE} \quad \checkmark$$

b) very rel. so  $E \gg E_0$   
 drop  $E_0$   $E \sim KE$  same approx result

$$\lambda \approx \frac{hc}{E} \approx \frac{hc}{KE}$$

So it looks like photons are as relativistic as things can get.

c) general  $p = \frac{1}{c} \sqrt{E^2 - E_0^2}$

$$= \frac{1}{c} \sqrt{(E - E_0)(E + E_0)}$$

$$= \frac{1}{c} \sqrt{KE (KE + 2E_0)}$$

$$\lambda = \frac{hc}{\sqrt{KE (KE + 2E_0)}}$$

We have been asked to write result in terms of KE and constants.  $E_0$  is a constant.

d) classical - so  $KE \ll E_0$

$$\lambda = \frac{hc}{\sqrt{KE (2E_0)}} \approx \frac{hc}{\sqrt{\frac{1}{2} m_0 v^2 (2 m_0 c^2)}} = \frac{h}{m_0 v}$$

↑ result

## 10-13

Calculate phase velocity of relativistic particle with rest mass greater than zero.

Phase velocity refers to speed of a wave with given frequency and wavelength---many ways to say that.

$$y = A \sin(kx - \omega t)$$

$$v_{\text{phase}} = \frac{\omega}{k} = \frac{\hbar\omega}{\hbar k} = \frac{h\nu}{h/\lambda} = \frac{E}{p}$$

$$\approx \frac{\sqrt{p^2 c^2 + E_0^2}}{p} = c \frac{\sqrt{p^2 + m_0^2 c^2}}{p}$$

So, for  $m_0 > 0$ , the phase velocity is greater than  $c$ . Note this does not mean any "thing" travels faster than light. Phase carries no energy, no information by itself.

10-14 Show that as group velocity tends to zero, the phase velocity becomes infinite.

We had

$$v_g = \frac{dE}{dp} = \frac{pc^2}{E} \quad \frac{d}{dt} \quad E^2 = p^2 c^2 + E_0^2$$

$$= \frac{c^2}{v_{\text{phase}}}$$

Clearly the phase velocity goes to infinity as group velocity goes to zero---except for zero rest mass particles where  $p=E/c$ . So a particle with zero velocity (or zero momentum) is "everywhere"---

11-1

What is the acceleration of an alpha particle at /near surface of gold nucleus.

$$m_{\alpha} = 6.7 \times 10^{-27} \text{ Kg}$$

From example 11.1 we have the electric field "at" the surface of the nucleus.

$$E = 2.0 \times 10^{21} \text{ N/C}$$

$$a = \frac{qE}{m} = 9.6 \times 10^{28} \text{ m/s}^2$$

That is a lot of "g's". There is no limit in relativity on acceleration.

11-6

Show that  $E(n) = -KE(n)$  in the Bohr model, and that  $KE(n) \ll \text{rest energy of the electron}$ .

$$PE = \frac{-ze^2}{4\pi\epsilon_0 r}$$

$$KE = \frac{ze^2}{8\pi\epsilon_0 r}$$

so

$$= -\frac{1}{2} PE$$

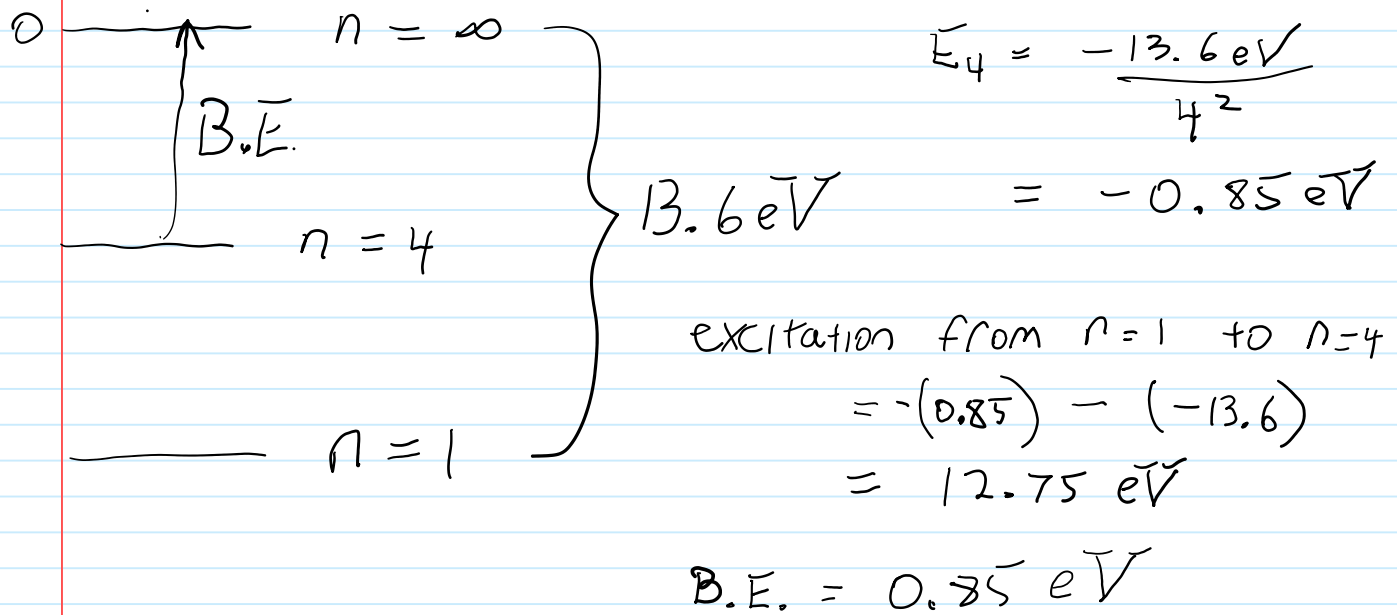
$$\begin{aligned} E &= PE + KE \\ &= PE + \left(-\frac{1}{2} PE\right) \\ &= +\frac{1}{2} PE \\ &= -KE \end{aligned}$$

For given  $n$   $|E_n|_{\max} = 13.6 \text{ eV} \ll 0.511 \text{ MeV}$

11.7 orbit for  $n = 120$

$$\begin{aligned} r_{120} &= (120)^2 r_0 \quad \leftarrow \begin{array}{l} 52.9 \text{ pm} \\ 52.9 \end{array} \\ &= 762 \text{ nm} - \underline{\underline{69}} \end{aligned}$$

11-9 What is binding energy for 3rd excited state (n=4 is 3rd excited state).



11-16 If the average lifetime of a state in Hydrogen (excited) is  $\sim 10^{-8}$  s, then what is energy width of that state and how many revolutions for orbit--in that time.

$$\Delta E = \frac{h}{\Delta t} = \frac{6.626 \times 10^{-34} \text{ J s}}{10^{-8} \text{ s}}$$

$$= 6.626 \times 10^{-26} \text{ J}$$

$$= 4.14 \times 10^{-7} \text{ eV}$$

Most states are near 13.6eV above ground state. The first excited state is 10.2 eV above ground! The energy width is small until one reaches a high n state ....when the states get close together, they may overlap (n+1) and n state.

$$b) \quad v = \frac{nh}{m(2\pi r_n)} = \frac{nh}{m(2\pi)(r_1 n^2)} = \frac{c}{137 n}$$

$$\# \text{ rev} = \frac{v \Delta t}{2\pi r_n} = \frac{c}{137 n} \left( \frac{10^{-8} \text{ s}}{2\pi n^2 r_1} \right)$$

$$\# \text{ rev} = \frac{1}{n^3} \frac{c}{137 \cdot 2\pi} \frac{10^{-2} \text{ s/Life}}{52.9 \times 10^{-12} \text{ m}}$$

$$= \frac{1}{n^3} 6.59 \times 10^7 \text{ rev/Lifetime}$$

High states go slow.